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CHAPTER 17 REVIEW Reaction
Kinetics MIXED REVIEW SHORT

ANSWER Answer the following
questions in the space provided. 1.

The reaction for the decomposition of
hydrogen peroxide is $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow$

$2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$. List three ways to

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speed up the rate of decomposition.
For each one, briefly explain why it is effective, based on collision theory.

17 Reaction Kinetics - David Brearley
High School

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Sarquis, Jerry L. Sarquis. Chapter 10
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Chapter Questions. 01:16. Problem 1
What idea is the kinetic-molecular

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theory based on? ... Problem 17
Explain the relationship between
atmospheric pressure and the actual
boiling point of a liquid. Rabia B.

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CHAPTER 12 & 13: SOLUTIONS The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc.

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