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Pure Substances and Mixtures, Elements \u0026 Compounds, Classification of Matter, Chemistry Examples, **States of Matter : Solid Liquid Gas** How does the stock market work? - Oliver Elfenbaum *Grit: the power of passion and perseverance* | Angela Lee Duckworth *What If We Could Open a Portal to a Parallel Universe?* *Dinosaur Pee?: Crash Course Kids #24.2 The Science of Lunch: Crash Course Kids #15.2 How Small Is An Atom? Spoiler: Very Small. What Are Pure Substances?* 3.2.1/3.2.2 Describe the differences between elements, compounds and mixtures. The Periodic Table: Atomic Radius, Ionization Energy, and Electronegativity Protein Synthesis (Updated) 3 States of Matter for Kids (Solid, Liquid, Gas): Science for Children - FreeSchool *Chemical Changes: Crash Course Kids #19.2 Finishing IT, Horror Books, and Another Book Haul | WEEKLY READING VLOG Chemistry 1.2 Classifying Matter (Part 1 of 3) **The Periodic Table: Crash Course Chemistry #4** Joeko Podcast 253: ~~The Ceiling You Can't Break Through is Made By You.~~ ~~With Dave Berke~~ ~~Introduction to Anatomy \u0026 Physiology: Crash Course A\u0026P #1~~ Biomolecules (Updated)*

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3) Identify and describe the particle in each phase of matter and how they are different in each phase of matter. ... Phases of Matter - Vocabulary Graphic Concepts Quiz. 1. Condensation. 2. Evaporation. 3. ... The iterative process of testing the most promising solutions and modifying what is proposed on the basis of the test.

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How would a chemist categorize each example of matter? saltwater; soil; water; oxygen; Answer a. Saltwater acts as if it were a single substance even though it contains two substances—salt and water. Saltwater is a homogeneous mixture, or a solution. Answer b. Soil is composed of small pieces of a variety of materials, so it is a heterogeneous mixture. Answer c

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Apply the kinetic theory to describe the motion of particles in a homogeneous mixture of sugar and a water as it is boiled As you heat something, molecules will start to move faster because the temperature increases. The molecules will spread further apart. As sugar water is heated the water will turn into vapor and leave the sugar behind.

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Physical properties describe the existence of matter, and chemical properties describe how substances change into other substances. A heterogeneous mixture is obviously a mixture, such as dirt; a homogeneous mixture behaves like a single substance, such as saltwater. solid to liquid (melting) and liquid to gas (boiling) (answers will vary)

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