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Nomenclature: Functional groups

Hess's Law

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-problems 1234 Practice Problemse 1. 2. 3. When a 12.8-g sample of KCl dissolves in 75.0 g of water in a calorimeter, the temperature drops from 31.00C to 21.60C. Calculate AH for the process. KCl(s) -> K+(aq) + Cl-(aq) even 6. What is the specific heat of gold if the temperature of a 8.21-g sample of gold is increased by 6.2 co when 6.51 J of heat

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Practice Problems 1. Identify each the following as a formula of an organic or inorganic compound: a. KCl b. C 4 H 10 c. C 2 H 6 O d. H 2 SO 4 e. CaCl 2 f. C 3 H 7 Cl 2. Identify each of the following as a formula of an organic or inorganic compound: a. C 6 H 12 O 6 b. K 3 PO 4 c. I 2 d. C 2 H 6 S e. C 10 H 22 f. C 4 H 9 Br 3.

Chapter 12 Introduction to Organic Chemistry: Hydrocarbons

Figure 12.1.4 Dissolution of 1 mol of an Ionic Compound In this case, dissolving 1 mol of (NH 4) 2 Cr 2 O 7 produces a solution that contains 1 mol of Cr 2 O 7 2- ions and 2 mol of NH 4 + ions. (Water molecules are omitted from a molecular view of the solution for clarity.)

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